

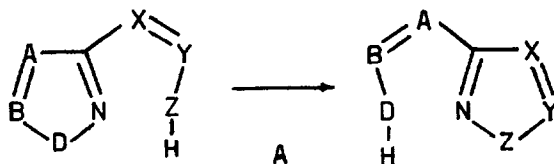
MONONUCLEAR HETEROCYCLIC REARRANGEMENTS - V.

1,2,4-OXADIAZOLES \longrightarrow IMIDAZOLES

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According to schema (A), proposed by Katritzky and coworkers¹ to generalize mononuclear heterocyclic rearrangements, a large range of transformations is available for different ABD and XYZ combinations. Nevertheless, all the examples in the literature^{1,2,3} are related to systems where D is always an oxygen atom and Z is always either an oxygen or a nitrogen atom.

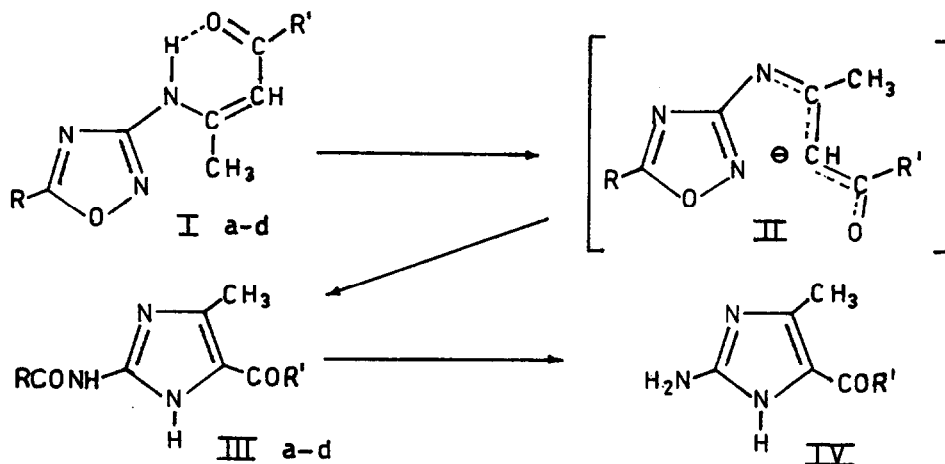


Thus, in connection with previous research in the field³, we have now investigated, within this type of reactions, the reaction behaviour of 1,2,4-oxadiazole enaminoketones Ia-d⁴. In fact, the NCC sequence⁵ (with terminal C potentially nucleophile) for these compounds, which are obtained by condensation between 3-amino-5-phenyl(methyl)-1,2,4-oxadiazoles and acetylacetone or benzoylacetone⁶, may show a reactivity analogous to that observed for sequences XYO and XYN, the only ones investigated so far.

Isomerization tests, carried out by heating (3 hrs. at 110°C) Ia-d with equimolar amounts of sodium ethoxide in *N,N*-dimethylformamide⁷, have allowed to ascertain the validity of this hypothesis. In fact, imidazole derivatives IIIa-d⁴ have been obtained with yield 60-80%. Their structures have been confirmed by acid hydrolysis which yielded benzoic or acetic acid and 2-aminoimidazoles IVa-b⁸.

The particular interest in the transformation of 1,2,4-oxadiazole enaminoketones Ia-d into imidazole derivatives IIIa-d depends on the fact that the latter represents the first example of mononuclear heterocyclic rearrangement (see schema A) where the new heterocycle closure occurs through the formation of C-N bond. The observed reaction is undoubtedly a simple and elegant way to synthesize certain imidazole derivatives.

Work in progress aims at elucidating the intrinsic mechanism of this rearrangement and at checking a possible generalization of the transformation realized, i.e. with $\underline{Z} = \underline{C}$, by changing the combination of both the starting heterocycle (\underline{ABD}) and the side chain (\underline{XYC}).



In I and III

- a: $R=C_6H_5$; $R'=CH_3$; b: $R=CH_3$; $R'=CH_3$
 c: $R=C_6H_5$; $R'=C_6H_5$; d: $R=CH_3$; $R'=C_6H_5$

In IV

- a: $R'=CH_3$
 b: $R'=C_6H_5$

REFERENCES

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2. Ref. 1 and references therein; K. Korbonits, *Ger.Offen.*, 2.038.919 (1969); *C.A.*, 74, 112039d (1971).
3. M. Ruccia and D. Spinelli, *Gazz.Chim.Ital.*, 89, 1654 (1959); M. Ruccia and N. Vivona, *Ann.Chim.(Italy)*, 57, 680 (1967); M. Ruccia and N. Vivona, *Chem.Comm.*, 866 (1970); M. Ruccia, N. Vivona, and G. Cusmano, *J.Heterocyclic Chem.*, 8, 137 (1971).
4. All the new compounds i.e., Ia (m.p. 112°C, ethanol), Ib (50°, petroleum ether), Ic (154°, ethanol), Id (125°, ethanol), IIIa (221°, ethanol/water 1:1), IIIb (233°, ethanol), IIIc (173°, ethanol/water 1:1), IIIId (170°, benzene) gave satisfactory analytical and nmr data.
5. Such a sequence in the side chain has been assumed by A.R. Katritzky (ref. 1).
6. Equimolar amounts are heated (8-10 hrs.) in anhydrous toluene, removing azeotropically the reaction water and using *p*-toluensulphonic acid as catalyzer.
7. Under different experimental conditions, a side reaction occurs when Ia-b are involved. Such reaction is now under study.
8. V. Grinsteins and A. Veneris, *Latvijas PSR Zinatnu Akad.Vestis, Kim.Ser.*, No. 3, 463 (1962); *C.A.* 59, 12784a (1963).